

X-Ray Diffraction Review Questions

Name _____

Date _____ Period _____

1. It is now possible to grow certain solids virtually an atomic layer at a time. Consider solid **A**, formed exclusively from copper and gold atoms, in which two atomic layers of gold atoms are put down, followed by two atomic layers of copper atoms, followed by two atomic layers of gold atoms, etc. The cycle is repeated until the desired thickness is reached, say, 100 atomic layers of each kind of atom; this is shown schematically below with X's and O's being Cu and Au, respectively.

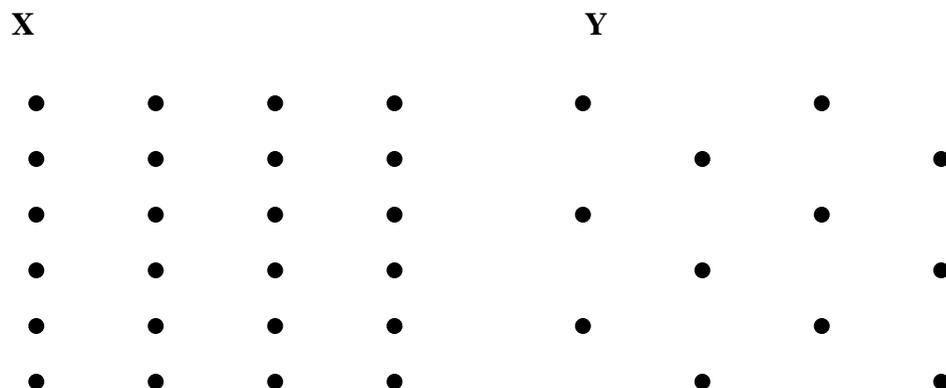
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XXXXXXXXXX
XXXXXXXXXX
OOOOOOOOO  cross-sectional view of solid A
OOOOOOOOO  X = Cu; O = Au
XXXXXXXXXX
XXXXXXXXXX
OOOOOOOOO
OOOOOOOOO etc.
```

In preparing solid **B**, in contrast, the growth is changed after each layer: an atomic layer of Cu is grown, then an atomic layer of Au, then another atomic layer of Cu, etc:

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XXXXXXXXXX
OOOOOOOOO  cross-sectional view of solid B
XXXXXXXXXX  X = Cu; O = Au
OOOOOOOOO etc.
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Why can diffraction be used to tell the two solids apart?

2. a. Infrared light and ultraviolet light can be used to generate a diffraction pattern from an array like those you worked with in lab. If your eyes were sensitive to infrared (IR) light with a wavelength of 8000\AA and to ultraviolet (UV) light with a wavelength of 3000\AA , using Fraunhofer's diffraction equation $m\lambda = d \sin \theta$ with d equal to the spacing between the lines in the diffraction grating (in this case $1.0 \times 10^6\text{\AA}$), find the values for θ and compare them.



- a. Based on your laboratory observations, which of the two diffraction patterns **X** and **Y** corresponds to array **a** and which to array **b**?
- b. Explain why having additional dots in the centers of the array of rectangles (array **b** relative to array **a**) has the effect you describe in part *a* on the diffraction pattern.
- c. Compare the relative sizes of unit cells for arrays **a** and **b**. What happens to the relative sizes of their diffraction patterns?
5. It has been predicted that if we could put the element hydrogen, normally an invisible gas of diatomic molecules at room temperature and pressure, under sufficiently high pressure – millions of atmospheres - it would become a metal!

If you could conduct a diffraction experiment on the hydrogen sample while it was being squeezed, what do you predict would happen to the spacing between diffraction spots as the atoms are placed under increasing pressure and why?

